



METHOD AND DEVICE FOR TRANSDERMAL  
ELECTROTRANSPORT DELIVERY OF FENTANYL  
AND SUFENTANIL

TECHNICAL FIELD

5 The invention relates generally to improved electrotransport drug delivery. Specifically, the invention relates to a device, composition and method for improved electrotransport delivery of analgesic drugs, particularly fentanyl and analogs of fentanyl. A composition is provided in the form of a hydrogel formulation for use in an electrotransport device.

BACKGROUND ART

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The transdermal delivery of drugs, by diffusion through the epidermis, offers improvements over more traditional delivery methods, such as subcutaneous injections and oral delivery. Transdermal drug delivery avoids the hepatic first pass effect encountered with oral drug delivery.

15 Transdermal drug delivery also eliminates patient discomfort associated with subcutaneous injections. In addition, transdermal delivery can provide more uniform concentrations of drug in the bloodstream of the patient over time due to the extended controlled delivery profiles of certain types of transdermal delivery devices. The term "transdermal" delivery, broadly

20 encompasses the delivery of an agent through a body surface, such as the skin, mucosa, or nails of an animal.

The skin functions as the primary barrier to the transdermal penetration of materials into the body and represents the body's major resistance to the transdermal delivery of therapeutic agents such as drugs.

25 To date, efforts have been focussed on reducing the physical resistance or

enhancing the permeability of the skin for the delivery of drugs by passive diffusion. Various methods for increasing the rate of transdermal drug flux have been attempted, most notably using chemical flux enhancers.

Other approaches to increase the rates of transdermal drug delivery include use of alternative energy sources such as electrical energy and ultrasonic energy. Electrically assisted transdermal delivery is also referred to as electrotransport. The term "electrotransport" as used herein refers generally to the delivery of an agent (eg, a drug) through a membrane, such as skin, mucous membrane, or nails. The delivery is induced or aided by application of an electrical potential. For example, a beneficial therapeutic agent may be introduced into the systemic circulation of a human body by electrotransport delivery through the skin. A widely used electrotransport process, electromigration (also called iontophoresis), involves the electrically induced transport of charged ions. Another type of electrotransport, electroosmosis, involves the flow of a liquid, which liquid contains the agent to be delivered, under the influence of an electric field. Still another type of electrotransport process, electroporation, involves the formation of transiently-existing pores in a biological membrane by the application of an electric field. An agent can be delivered through the pores either passively (ie, without electrical assistance) or actively (ie, under the influence of an electric potential). However, in any given electrotransport process, more than one of these processes, including at least some "passive" diffusion, may be occurring simultaneously to a certain extent. Accordingly, the term "electrotransport", as used herein, should be given its broadest possible interpretation so that it includes the electrically induced or enhanced transport of at least one agent, which may be charged, uncharged, or a mixture thereof, whatever the specific mechanism or mechanisms by which the agent actually is transported.

Electrotransport devices use at least two electrodes that are in electrical contact with some portion of the skin, nails, mucous membrane, or other surface of the body. One electrode, commonly called the "donor" electrode, is the electrode from which the agent is delivered into the body. 5 The other electrode, typically termed the "counter" electrode, serves to close the electrical circuit through the body. For example, if the agent to be delivered is positively charged, ie, a cation, then the anode is the donor electrode, while the cathode is the counter electrode which serves to complete the circuit. Alternatively, if an agent is negatively charged, ie, an 10 anion, the cathode is the donor electrode and the anode is the counter electrode. Additionally, both the anode and cathode may be considered donor electrodes if both anionic and cationic agent ions, or if uncharged dissolved agents, are to be delivered.

Furthermore, electrotransport delivery systems generally require at 15 least one reservoir or source of the agent to be delivered to the body. Examples of such donor reservoirs include a pouch or cavity, a porous sponge or pad, and a hydrophilic polymer or a gel matrix. Such donor reservoirs are electrically connected to, and positioned between, the anode or cathode and the body surface, to provide a fixed or renewable source of 20 one or more agents or drugs. Electrotransport devices also have an electrical power source such as one or more batteries. Typically at any one time, one pole of the power source is electrically connected to the donor electrode, while the opposite pole is electrically connected to the counter electrode. Since it has been shown that the rate of electrotransport drug 25 delivery is approximately proportional to the electric current applied by the device, many electrotransport devices typically have an electrical controller that controls the voltage and/or current applied through the electrodes, thereby regulating the rate of drug delivery. These control circuits use a variety of electrical components to control the amplitude, polarity, timing,

waveform shape, etc. of the electric current and/or voltage supplied by the power source. See, for example, McNichols et al., U.S. Patent 5,047,007.

To date, commercial transdermal electrotransport drug delivery devices (eg, the Phoresor, sold by Iomed, Inc. of Salt Lake City, UT; the Dupel Iontophoresis System sold by Empi, Inc. of St. Paul, MN; the Webster Sweat Inducer, model 3600, sold by Wescor, Inc. of Logan, UT) have generally utilized a desk-top electrical power supply unit and a pair of skin contacting electrodes. The donor electrode contains a drug solution while the counter electrode contains a solution of a biocompatible electrolyte salt. The power supply unit has electrical controls for adjusting the amount of electrical current applied through the electrodes. The "satellite" electrodes are connected to the electrical power supply unit by long (eg, 1-2 meters) electrically conductive wires or cables. The wire connections are subject to disconnection and limit the patient's movement and mobility. Wires between electrodes and controls may also be annoying or uncomfortable to the patient. Other examples of desk-top electrical power supply units which use "satellite" electrode assemblies are disclosed in Jacobsen et al., U.S. Patent 4,141,359 (see Figures 3 and 4); LaPrade, U.S. Patent 5,006,108 (see Figure 9); and Maurer et al., U.S. Patent 5,254,081.

More recently, small self-contained electrotransport delivery devices have been proposed to be worn on the skin, sometimes unobtrusively under clothing, for extended periods of time. Such small self-contained electrotransport delivery devices are disclosed for example in Tapper, U.S. Patent 5,224,927; Sibalis, et al., U.S. Patent 5,224,928; and Haynes et al., U.S. Patent 5,246,418.

There have recently been suggestions to utilize electrotransport devices having a reusable controller which is adapted for use with multiple drug-containing units. The drug-containing units are simply disconnected

from the controller when the drug becomes depleted and a fresh drug-containing unit is thereafter connected to the controller. In this way, the relatively more expensive hardware components of the device (eg, batteries, LED's, circuit hardware, etc.) can be contained within the reusable controller, and the relatively less expensive donor reservoir and counter reservoir matrices can be contained in the single use/disposable drug-containing unit, thereby bringing down the overall cost of electrotransport drug delivery. Examples of electrotransport devices comprised of a reusable controller, removably connected to a drug-containing unit are disclosed in Sage, Jr. et al., U.S. Patent 5,320,597; Sibalis, U.S. Patent 5,358,483; Sibalis et al., U.S. Patent 5,135,479 (Fig. 12); and Devane et al., UK Patent Application 2 239 803.

In further development of electrotransport devices, hydrogels have become particularly favored for use as the drug and electrolyte reservoir matrices, in part, due to the fact that water is the preferred liquid solvent for use in electrotransport drug delivery due to its excellent biocompatibility compared with other liquid solvents such as alcohols and glycols. Hydrogels have a high equilibrium water content and can quickly absorb water. In addition, hydrogels tend to have good biocompatibility with the skin and with mucosal membranes.

Of particular interest in transdermal delivery is the delivery of analgesic drugs for the management of moderate to severe pain. Control of the rate and duration of drug delivery is particularly important for transdermal delivery of analgesic drugs to avoid the potential risk of overdose and the discomfort of an insufficient dosage.

One class of analgesics that has found application in a transdermal delivery route is the synthetic opiates, a group of 4-aniline piperidines. The synthetic opiates, eg, fentanyl and certain of its derivatives such as

sufentanil, are particularly well-suited for transdermal administration. These synthetic opiates are characterized by their rapid onset of analgesia, high potency, and short duration of action. They are estimated to be 80 and 800 times, respectively, more potent than morphine. These drugs are weak bases, ie, amines, whose major fraction is cationic in acidic media.

In an *in vivo* study to determine plasma concentration, Thysman and Preat (*Anesth. Analg.* 77 (1993) pp. 61-66) compared simple diffusion of fentanyl and sufentanil to electrotransport delivery in citrate buffer at pH 5. Simple diffusion did not produce any detectable plasma concentration. The plasma levels attainable depended on the maximum flux of the drug that can cross the skin and the drug's pharmacokinetic properties, such as clearance and volume of distribution. Electrotransport delivery was reported to have significantly reduced lag time (ie, time required to achieve peak plasma levels) as compared to passive transdermal patches (1.5 h versus 14 h). The researchers' conclusions were that electrotransport of these analgesic drugs can provide more rapid control of pain than classical patches, and a pulsed release of drug (by controlling electrical current) was comparable to the constant delivery of classical patches. See, also, eg, Thysman et al. *Int. J. Pharma.*, 101 (1994) pp. 105-113; V. Pr  at et al. *Int. J. Pharm.*, 96 (1993) pp. 189-196 (sufentanil); G  urlav et al. *Pain*, 37 (1989) pp. 193-202 (fentanyl); Sebel et al. *Eur. J. Clin. Pharmacol.* 32 (1987) pp. 529-531 (fentanyl and sufentanil). Passive, ie, by diffusion, and electrically-assisted transdermal delivery of narcotic analgesic drugs, such as fentanyl, to induce analgesia, have also both been described in the patent literature. See, for example, Gale et al., U.S. Patent 4,588,580, and Theeuwes et al., U.S. Patent 5,232,438.

In the last several years, management of post-operative pain has looked to delivery systems other than electrotransport delivery. Particular attention has been given to devices and systems which permit, within

predetermined limits, the patient to control the amount of analgesic the patient receives. The experience with these types of devices has generally been that patient control of the administration of analgesic has resulted in the administration of less analgesic to the patient than would have been administered were the dosage prescribed by a physician. Self-administered  
5 or patient controlled self-administration has become known (and will be referred to herein) as patient-controlled analgesia (PCA).

Known PCA devices are typically electromechanical pumps which require large capacity electrical power sources, eg, alternating current or  
10 multiple large capacity battery packs which are bulky. Due to their bulk and complexity, commercially available PCA devices generally require the patient to be confined to a bed, or some other essentially fixed location. Known PCA devices deliver drug to the patient by means of an intravenous line or a catheter which must be inserted into the intended vein, artery or other organ  
15 by a qualified medical technician. This technique requires that the skin barrier be breached in order to administer the analgesic. (See, Zdeb U.S. Patent 5,232,448). Thus, as practiced using commercially available PCA devices, PCA requires the presence of highly skilled medical technicians to initiate and supervise the operation of the PCA device along with its  
20 attendant risk of infection. Further, commercially available PCA devices themselves are somewhat painful to use by virtue of their percutaneous (ie, intravenous or subcutaneous) access.

The art has produced little in the way of transdermal electrotransport devices that can compete with the conventional PCAs in terms of the  
25 amount of drug delivered to achieve adequate analgesia and in a patient controlled manner. Further, little progress has been made to provide a hydrogel formulation for analgesic electrotransport, particularly fentanyl transdermal electrotransport delivery, that has long term stability and has performance characteristics comparable to the patient controlled

electromechanical pumps for, eg, intravenous delivery of analgesic. There is need to provide an analgesic formulation in a suitable device to take advantage of the convenience of electrotransport delivery in a small, self-contained, patient-controlled device.

## DESCRIPTION OF THE INVENTION

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The present invention provides a method for improved transdermal electrotransport delivery of fentanyl and analogs of fentanyl, particularly sufentanil. As such, the method of the present invention provides a greater degree of efficiency in electrotransport delivery of analgesic fentanyl or sufentanil, concomitantly providing a greater measure of patient safety and comfort in pain management. The foregoing, and other advantages of the present invention, are provided by a method of delivering fentanyl or sufentanil through a body surface (eg, intact skin) by electrotransport from an electrotransport delivery device having a anodic donor reservoir containing an at least partially aqueous solution of a fentanyl/sufentanil salt.

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In one aspect, the invention concerns maintaining the concentration of fentanyl or sufentanil salt in the donor reservoir solution at or above a level at which the transdermal fentanyl or sufentanil flux begins to become dependent on the concentration of the drug in solution. For fentanyl, the transdermal electrotransport flux remains independent of fentanyl concentration at or above about 11 to 16 mM substantially throughout the fentanyl electrotransport delivery period. By maintaining the concentration of fentanyl salt solution at or above about 11 to 16 mM in the donor reservoir, the electrotransport flux of the drug remains substantially independent of the drug concentration in the donor reservoir solution and substantially proportional to the level of electrotransport current applied by the delivery device during the electrotransport drug delivery. Maintaining the fentanyl salt solution concentration above about 11 mM, and preferably above about 16

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mM ensures a predictable fentanyl flux with a particular applied electrotransport current.

For sufentanil, the transdermal electrotransport flux remains independent of sufentanil concentration at or above about 1.7 mM substantially throughout the sufentanil electrotransport delivery period. By  
5 maintaining the concentration of sufentanil salt solution at or above about 1.7 mM in the donor reservoir, the electrotransport flux of the drug remains substantially independent of the drug concentration in the donor reservoir solution and substantially proportional to the level of electrotransport current  
10 applied by the delivery device during the electrotransport drug delivery. Maintaining the sufentanil salt solution concentration above about 1.7 mM ensures a predictable sufentanil flux with a particular applied electrotransport current.

In another aspect, the invention provides a donor reservoir formulation  
15 for a transdermal electrotransport fentanyl/sufentanil delivery device having an anodic donor electrode comprised of silver, which donor reservoir formulation substantially prevents migration of silver ions into, and discoloration of, the skin of the patient. While the prior art has taught the advantage of using a halide drug salt to prevent the migration of  
20 electrochemically generated silver ions (see Untereker et al U.S. Patent 5,135,477), it has now been discovered that for halide salts of fentanyl or sufentanil which are delivered either continuously or intermittently over longer electrotransport delivery periods (eg, periods of at least several hours), the amount of fentanyl/sufentanil halide needed in the donor reservoir in order to  
25 prevent this silver migration must be well in excess of the amount of fentanyl/sufentanil which is needed for therapeutic purposes. For fentanyl hydrochloride, the amount of drug needed to prevent silver ion migration has been determined to be at least about 3 times the amount needed for delivery into the patient at least under the specific electrotransport delivery

conditions (ie, applied electrotransport current, reservoir size/weight/composition, and time of electrotransport current application) which are described in more detail hereinafter.

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5 In yet another aspect, the present invention concerns a method of administering fentanyl or <sup>sufentanil</sup> ~~sufentanil~~ by transdermal electrotransport in order to treat moderate-to-severe pain associated with major surgical procedures. We have determined that a transdermal electrotransport dose of about 20  $\mu$ g to about 60  $\mu$ g of fentanyl, delivered over a delivery interval of up to about 20 minutes, is therapeutically effective in treating moderate-to-severe post-operative pain in human patients having body weights above about 35 kg. 10 Preferably, the amount of fentanyl delivered is about 35  $\mu$ g to about 45  $\mu$ g over a delivery interval of about 5 to 15 minutes, and most preferably the amount of fentanyl delivered is about 40  $\mu$ g over a delivery interval of about 10 minutes. Since fentanyl has a relatively short distribution half life once 15 delivered into a human body (ie, about 3 hours), the method of inducing analgesia preferably includes a method for maintaining the analgesia so induced. Thus the method of transdermally delivering fentanyl by electrotransport preferably includes delivering at least 1 additional, more preferably about 10 to 100 additional, and most preferably about 20 to 80 20 additional, like dose(s) of fentanyl over subsequent like delivery interval(s) over a 24 hour period. The ability to deliver multiple identical doses from a transdermal electrotransport fentanyl delivery device also provides the capability of pain management to a wider patient population, in which different patients require different amounts of fentanyl to control their pain. 25 By providing the capability of administering multiple small transdermal electrotransport fentanyl doses, the patients can titrate themselves to administer only that amount of fentanyl which is needed to control their pain, and no more.

Other advantages and a fuller appreciation of specific adaptations, compositional variations, and physical attributes of the present invention can be learned from an examination of the following drawings, detailed description, examples, and appended claims.

## BRIEF DESCRIPTION OF THE DRAWINGS

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The present invention is hereinafter described in conjunction with the appended drawings, in which like designations refer to like elements throughout, and in which:

Figure 1 is a perspective exploded view of an electrotransport drug  
10 delivery device in accordance with the present invention;

Figure 2 is a graph of normalized transdermal electrotransport flux versus concentration of fentanyl HCl in aqueous solution;

Figure 3 is a graph illustrating quality of analgesia in patients administered with transdermal electrotransport fentanyl as a function of time;  
15 and

Figure 4 is a graph illustrating pain intensity experienced by patients administered transdermal electrotransport fentanyl as a function of time.

## MODES FOR CARRYING OUT THE INVENTION

The present invention relates broadly to improved methods for the  
20 transdermal electrotransport delivery of fentanyl or sufentanil, in water soluble salt form, in formulations to achieve a systemic analgesic effect.

## MINIMUM FENTANYL/SUFENTANIL CONCENTRATION TO MAINTAIN PREDICTABLE ELECTROTRANSPORT FLUX

In one aspect, the present invention is characterized by maintaining the transdermal electrotransport fentanyl/sufentanil flux independent of drug concentration in the donor reservoir during the electrotransport drug delivery period. In another aspect, the present invention concerns a fentanyl or sufentanil halide donor reservoir composition, which is adapted to be used in an electrotransport delivery device having a silver anodic donor electrode, which formulation is effective to prevent skin discoloration from silver ions formed during oxidation of the silver anode. In yet another aspect, the present invention provides a fentanyl or sufentanil salt electrotransport delivery device, and a method of using same, to achieve an analgesic effect which is comparable to the effect achieved in known IV accessed patient controlled analgesic pumps.

Concerning the first aspect of the present invention, transdermal electrotransport fentanyl flux begins to become dependent upon the concentration of the fentanyl salt in aqueous solution as the fentanyl salt concentration falls below about 11 to 16 mM. The 11 to 16 mM concentration is calculated based only on the volume of liquid solvent used in the donor reservoir, not on the total volume of the reservoir. In other words, the 11 to 16 mM concentration does not include the volume of the reservoir which is represented by the reservoir matrix (eg, hydrogel or other matrix) material. Furthermore, the 11 to 16 mM concentration is based upon the number of moles of fentanyl salt, not the equivalent number of moles of fentanyl free base, which is contained in the donor reservoir solution.

For fentanyl HCl, the 11 to 16 mM concentration is equivalent to about 4 to 6 mg/mL. Other fentanyl salts (eg, fentanyl citrate) will have slightly differing weight based concentration ranges based on the difference

in the molecular weight of the counter ion of the particular fentanyl salt in question.

As the fentanyl salt concentration falls to about 11 to 16 mM, the fentanyl transdermal electrotransport flux begins to significantly decline, even  
5 if the applied electrotransport current remains constant. Thus, to ensure a predictable fentanyl flux with a particular level of applied electrotransport current, the fentanyl salt concentration in the solution contained in the donor reservoir should be maintained above about 11 mM, and preferably above about 16 mM. This aspect of the present invention maintains the fentanyl  
10 salt concentration in solution above a minimum level to ensure a predictable transdermal electrotransport flux at any particular applied electrotransport current level.

In addition to fentanyl, water soluble salts of sufentanil also have minimum aqueous solution concentrations below which the transdermal  
15 electrotransport flux becomes dependent on concentration of the sufentanil salt in solution. The minimum concentration for sufentanil is about 1.7 mM, which for sufentanil citrate is equivalent to about 1 mg/mL.

As long as there is no binding of the fentanyl/sufentanil to the reservoir matrix material, the particular matrix material chosen as the donor  
20 reservoir matrix has little if any effect on the minimum concentration needed to assure predictable transdermal electrotransport fentanyl/sufentanil flux. Hydrogel matrices in particular exhibit no tendency to bind fentanyl or sufentanil and so hydrogels are a preferred class of matrix materials for use with this aspect of the present invention.

PREVENTION OF SILVER MIGRATION DURING  
FENTANYL/SUFENTANIL ELECTROTRANSPORT

The second aspect of the invention concerns the delivery of fentanyl or sufentanil from an electrotransport device having a silver donor electrode.

5 Since fentanyl and sufentanil are both bases, the salts of fentanyl and sufentanil are typically acid addition salts, eg, citrate salts, hydrochloride salts, etc. The acid addition salts of fentanyl typically have water solubilities of about 25 to 30 mg/mL. The acid addition salts of sufentanil typically have water solubilities of about 45 to 50 mg/mL. When these salts are placed in

10 solution (eg, aqueous solution), the salts dissolve and form protonated fentanyl or sufentanil cations and counter (eg, citrate or chloride) anions. As such, the fentanyl/sufentanil cations are delivered from the anodic electrode of an electrotransport delivery device. Silver anodic electrodes have been proposed for transdermal electrotransport delivery as a way to maintain pH

15 stability in the anodic reservoir. See for example, Untereker et al U.S. Patent 5,135,477 and Petelenz et al U.S. Patent 4,752,285. These patents also recognize one of the shortcomings of using a silver anodic electrode in an electrotransport delivery device, namely that the application of current through the silver anode causes the silver to become oxidized ( $\text{Ag} \rightarrow \text{Ag}^+ + \text{e}^-$ )

20 ) thereby forming silver cations which compete with the cationic drug for delivery into the skin by electrotransport. Silver ion migration into the skin results in a transient epidermal discoloration (TED) of the skin. In addition to these patents, Phipps et al PCT/US95/04497 filed on April 7, 1995 teaches the use of supplementary chloride ion sources in the form of high molecular

25 weight chloride resins in the donor reservoir of a transdermal electrotransport delivery device. While these resins are highly effective at providing sufficient chloride for preventing silver ion migration, and the attendant skin discoloration, these resins can also have adverse reactions with either the drug being delivered (ie, binding of drug to the resin) and/or with the skin of

30 the patient (ie, contributing to skin irritation reactions). Thus, for the

purposes of the following discussion, the donor reservoir formulations of the present invention will be assumed to be substantially free of such secondary chloride ion source resins. While the Untereker and Petelenz patents teach that providing a cationic drug in the form of a halide salt prevents the migration of silver ions (ie, by reacting the silver ions with the halide counter ion of the drug to form a water insoluble silver halide precipitate;  $\text{Ag}^+ + \text{X}^- \rightarrow \text{AgX}$ ), we have surprisingly discovered that a significant excess (ie, an amount well in excess of the fentanyl halide salt needed to be delivered to the patient for purposes of achieving analgesia) of fentanyl halide must be provided in a donor reservoir of an electrotransport fentanyl delivery device in order to prevent silver ion migration. This is especially true for those transdermal electrotransport delivery devices which are adapted to apply electrotransport current for extended periods of time, eg, longer than about 6 hours.

In general, the "excess" amount of fentanyl halide needed to prevent silver ion migration will be highly dependent upon a number of factors including the particular halide salt used (eg, chloride, fluoride, bromide or iodide salt of the drug), the level of applied electrotransport current, the size/weight/composition of the donor reservoir, the applied current density level and the length of time over which the electrotransport current is applied. We have determined delivering fentanyl hydrochloride from polyvinyl alcohol based donor reservoirs which are used to deliver fentanyl for periods of up to about 15 hours, that the amount of fentanyl HCl needed to prevent silver ion migration during electrotransport delivery is about 2 to 3 times the amount of fentanyl HCl needed for delivery into the patient over that same period of time for purposes of inducing and maintaining analgesia.

In the specific case of an electrotransport delivery device having a polyvinyl alcohol based donor reservoir containing fentanyl hydrochloride and having a total weight (on a hydrated basis) of about 0.3 to 0.8 g, which

device (1) has an anodic donor electrode comprised of silver (eg, silver foil or silver powder-loaded polymer film) which is in electrical contact with the donor reservoir, (2) has an electrical power source which applies a DC current of about 190  $\mu$ A to 230  $\mu$ A to the donor and counter electrodes, (3) applies a current density, measured as the total applied current divided by the skin contact area of the donor reservoir, of less than about 0.3 mA/cm<sup>2</sup>, and (4) is capable of applying such current for up to about eighty separate delivery intervals of about 8 to about 12 minutes duration, the fentanyl HCl loading needed to induce and maintain analgesia is about 2.5 to 3.5 mg, yet the fentanyl HCl loading needed to prevent TED is at least about 8 to 10 mg, and preferably at least about 11 to 13 mg. More specifically in the case of an electrotransport delivery device having a polyvinyl alcohol based donor reservoir containing fentanyl hydrochloride and having a total weight (on a hydrated basis) of about 0.5 to 0.8 g, which device applies a DC current of about 210  $\mu$ A to the electrodes, and is capable of applying such current for up to about eighty separate delivery intervals of about 10 minutes duration, the fentanyl HCl loading needed to induce and maintain analgesia is about 3 mg, yet the fentanyl HCl loading needed to prevent TED is at least about 9 mg, and preferably at least about 12 mg.

In order to determine the loading of a halide salt of fentanyl other than fentanyl HCl, it is only necessary to supply an equivalent molar amount of halide ions to the reservoir since the silver halide salts have fairly uniformly low water solubility. For example, the loading of 8 to 10 mg of fentanyl HCl corresponds to a molar loading of about 20 to 25  $\mu$ moles. Thus, about 20 to 25  $\mu$ moles of any of the other fentanyl halides (ie, fentanyl fluoride, fentanyl bromide or fentanyl iodide) will provide an equivalent degree of silver migration prevention as fentanyl HCl.

In addition to fentanyl, "excess" amounts of sufentanil halide salts also can be used to prevent silver ion migration. Because sufentanil is



about 7 to 10 times more potent than fentanyl, only about 0.1 to 0.14 times the fentanyl dose is needed to achieve an equivalent level of analgesia. However, because the transdermal electrotransport delivery efficiency of sufentanil (ie, the rate of sufentanil delivered per unit of applied electrotransport current) is only about one-third that of fentanyl, the applied electrotransport current needed to achieve the same level of analgesia with sufentanil is about 0.3 to 0.4 times that needed for fentanyl. Thus, the "excess" amount of sufentanil chloride needed to prevent silver ion migration during electrotransport delivery of sufentanil is correspondingly reduced to about 6 to 10  $\mu$ moles or about 2.4 to 4 mg. The amount of sufentanil HCl loading needed to prevent silver ion migration, relative to the loading needed to achieve an analgesic effect in a patient, is at least about 4 times the analgesically effective loading.

As long as the reservoir matrix material has substantially no silver ion binding capacity (ie, by means of a fixed anionic (eg,  $\text{COO}^-$ ) moiety as is found in cation exchange membranes), the particular matrix material chosen as the donor reservoir matrix has little if any effect on the minimum loading of halide salts of fentanyl and sufentanil which is effective to prevent silver ion migration into the patient's skin. Hydrogel matrices in particular exhibit little or no tendency to bind silver ions and so are a preferred matrix material for use with this aspect of the present invention.

#### TRANSDERMAL ELECTROTRANSPORT FENTANYL/SUFENTANIL DOSING FOR INDUCING AND MAINTAINING ANALGESIA

In another aspect, the present invention provides a method and electrotransport delivery device for delivering fentanyl or <sup>sufentanil</sup> ~~sufentanil~~ through a body surface, eg, skin, to achieve an analgesic effect. The fentanyl or

sufentanil salt is provided in a donor reservoir of an electrotransport delivery device as an aqueous salt solution.

The dose of fentanyl delivered by transdermal electrotransport is  
5 about 20  $\mu\text{g}$  to about 60  $\mu\text{g}$  over a delivery time of up to about 20 minutes in human patients having body weights of 35 kg or greater. Preferred is a dosage of about 35  $\mu\text{g}$  to about 45  $\mu\text{g}$ , and most preferred is a dosage of about 40  $\mu\text{g}$  for the delivery period. The method of the invention further preferably includes delivery of about 10 to 100, and more preferably about  
10 20 to 80 additional like doses over a period of 24 hours in order to achieve and maintain the analgesic effect.

The dose of sufentanil delivered by transdermal electrotransport is about 2.3  $\mu\text{g}$  to about 7.0  $\mu\text{g}$  over a delivery time of up to about 20 minutes in human patients having a body weights of 35 kg or greater. Preferred is a  
15 dosage of about 4  $\mu\text{g}$  to about 5.5  $\mu\text{g}$ , and most preferred is a dosage of about 4.7  $\mu\text{g}$  for the delivery period. The method of the invention further preferably includes delivery of about 10 to 100, and more preferably about 20 to 80 additional like doses over a period of 24 hours in order to achieve and maintain the analgesic effect.

20 The fentanyl/sufentanil salt-containing anodic reservoir formulation for transdermally delivering the above mentioned doses of fentanyl/sufentanil by electrotransport is preferably comprised of an aqueous solution of a water soluble fentanyl/sufentanil salt such as HCl or citrate salts. Most preferably, the aqueous solution is contained within a hydrophilic polymer matrix such  
25 as a hydrogel matrix. The fentanyl/sufentanil salt is present in an amount sufficient to deliver the above mentioned doses transdermally by electrotransport over a delivery period of up to about 20 minutes, to achieve a systemic analgesic effect. The fentanyl/sufentanil salt typically comprises about 1 to 10 wt% of the donor reservoir formulation (including the weight of

the polymeric matix) on a fully hydrated basis, and more preferably about 1 to 5 wt% of the donor reservoir formulation on a fully hydrated basis.

Although not critical to this aspect of the present invention, the applied electrotransport current density is typically in the range of about 50 to 150  $\mu\text{A}/\text{cm}^2$  and the applied electrotransport current is typically in the range of about 150 to 240  $\mu\text{A}$ .

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The anodic fentanyl/sufentanil salt-containing hydrogel can suitably be made of ~~a~~ any number of materials but preferably is comprised of a hydrophilic polymeric material, preferably one that is polar in nature so as to enhance the drug stability. Suitable polar polymers for the hydrogel matrix comprise a variety of synthetic and naturally occurring polymeric materials. A preferred hydrogel formulation contains a suitable hydrophilic polymer, a buffer, a humectant, a thickener, water and a water soluble fentanyl or sufentanil salt (eg, HCl salt). A preferred hydrophilic polymer matrix is polyvinyl alcohol such as a washed and fully hydrolyzed polyvinyl alcohol (PVOH), eg, Mowiol 66-100 commercially available from Hoechst Aktiengesellschaft. A suitable buffer is an ion exchange resin which is a copolymer of methacrylic acid and divinylbenzene in both an acid and salt form. One example of such a buffer is a mixture of Polacrillin (the copolymer of methacrylic acid and divinyl benzene available from Rohm & Haas, Philadelphia, PA) and the potassium salt thereof. A mixture of the acid and potassium salt forms of Polacrillin functions as a polymeric buffer to adjust the pH of the hydrogel to about pH 6. Use of a humectant in the hydrogel formulation is beneficial to inhibit the loss of moisture from the hydrogel. An example of a suitable humectant is guar gum. Thickeners are also beneficial in a hydrogel formulation. For example, a polyvinyl alcohol thickener such as hydroxypropyl methylcellulose (eg, Methocel K100MP available from Dow Chemical, Midland, MI) aids in modifying the rheology of a hot polymer solution as it is dispensed into a mold or cavity. The hydroxypropyl

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methycellulose increases in viscosity on cooling and significantly reduces the propensity of a cooled polymer solution to overfill the mold or cavity.

a In one preferred embodiment, the anodic ~~fen~~<sup>fentanyl/sufentanil</sup> ~~tanyl/sufentanil~~ salt-containing hydrogel formulation comprises about 10 to 15 wt% polyvinyl alcohol, 0.1 to 0.4 wt% resin buffer, and about 1 to 2 wt% fentanyl or sufentanil salt, preferably the hydrochloride salt. The remainder is water and ingredients such as humectants, thickeners, etc. The polyvinyl alcohol (PVOH)-based hydrogel formulation is prepared by mixing all materials, including the fentanyl or sufentanil salt, in a single vessel at elevated temperatures of about 90°C to 95°C for at least about 0.5 hr. The hot mix is then poured into foam molds and stored at freezing temperature of about -35°C overnight to cross-link the PVOH. Upon warming to ambient temperature, a tough elastomeric gel is obtained suitable for fentanyl electrotransport.

15 The hydrogel formulations are used in an electrotransport device such as described hereinafter. A suitable electrotransport device includes an anodic donor electrode, preferably comprised of silver, and a cathodic counter electrode, preferably comprised of silver chloride. The donor electrode is in electrical contact with the donor reservoir containing the aqueous solution of a fentanyl/sufentanil salt. As described above, the donor reservoir is preferably a hydrogel formulation. The counter reservoir also preferably comprises a hydrogel formulation containing a (eg, aqueous) solution of a biocompatible electrolyte, such as citrate buffered saline. The anodic and cathodic hydrogel reservoirs preferably each have a skin contact area of about 1 to 5 cm<sup>2</sup> and more preferably about 2 to 3 cm<sup>2</sup>. The anodic and cathodic hydrogel reservoirs preferably have a thickness of about 0.05 to 0.25 cm, and more preferably about 0.15 cm. The applied electrotransport current is about 150 µA to about 240 µA, depending on the

analgesic effect desired. Most preferably, the applied electrotransport current is substantially constant DC current during the dosing interval.

Reference is now made to FIG. 1 which depicts an exemplary electrotransport device which can be used in accordance with the present invention. FIG. 1 shows a perspective exploded view of an electrotransport device 10 having an activation switch in the form of a push button switch 12 and a display in the form of a light emitting diode (LED) 14. Device 10 comprises an upper housing 16, a circuit board assembly 18, a lower housing 20, anode electrode 22, cathode electrode 24, anode reservoir 26, cathode reservoir 28 and skin-compatible adhesive 30. Upper housing 16 has lateral wings 15 which assist in holding device 10 on a patient's skin. Upper housing 16 is preferably composed of an injection moldable elastomer (eg, ethylene vinyl acetate). Printed circuit board assembly 18 comprises an integrated circuit 19 coupled to discrete electrical components 40 and battery 32. Circuit board assembly 18 is attached to housing 16 by posts (not shown in FIG. 1) passing through openings 13a and 13b, the ends of the posts being heated/melted in order to heat stake the circuit board assembly 18 to the housing 16. Lower housing 20 is attached to the upper housing 16 by means of adhesive 30, the upper surface 34 of adhesive 30 being adhered to both lower housing 20 and upper housing 16 including the bottom surfaces of wings 15.

Shown (partially) on the underside of circuit board assembly 18 is a battery 32, which is preferably a button cell battery and most preferably a lithium cell. Other types of batteries may also be employed to power device 10.

The circuit outputs (not shown in FIG. 1) of the circuit board assembly 18 make electrical contact with the electrodes 24 and 22 through openings 23,23' in the depressions 25,25' formed in lower housing, by means of electrically conductive adhesive strips 42,42'. Electrodes 22 and 24, in turn,

are in direct mechanical and electrical contact with the top sides 44',44 of drug reservoirs 26 and 28. The bottom sides 46',46 of drug reservoirs 26,28 contact the patient's skin through the openings 29',29 in adhesive 30. Upon depression of push button switch 12, the electronic circuitry on circuit board assembly 18 delivers a predetermined DC current to the  
5 electrodes/reservoirs 22,26 and 24,28 for a delivery interval of predetermined length, eg, about 10 minutes. Preferably, the device transmits to the user a visual and/or audible confirmation of the onset of the drug delivery, or bolus, interval by means of LED 14 becoming lit and/or an audible sound signal  
10 from, eg, a "beeper". Analgesic drug, eg fentanyl, is then delivered through the patient's skin, eg, on the arm, for the predetermined (eg, 10 minute) delivery interval. In practice, a user receives feedback as to the onset of the drug delivery interval by visual (LED 14 becomes lit) and/or audible signals (a beep from the "beeper").

15 Anodic electrode 22 is preferably comprised of silver and cathodic electrode 24 is preferably comprised of silver chloride. Both reservoirs 26 and 28 are preferably comprised of polymer hydrogel materials as described herein. Electrodes 22, 24 and reservoirs 26, 28 are retained by lower housing 20. For fentanyl and sufentanil salts, the anodic reservoir 26 is the  
20 "donor" reservoir which contains the drug and the cathodic reservoir 28 contains a biocompatible electrolyte.

The push button switch 12, the electronic circuitry on circuit board assembly 18 and the battery 32 are adhesively "sealed" between upper housing 16 and lower housing 20. Upper housing 16 is preferably  
25 composed of rubber or other elastomeric material. Lower housing 20 is preferably composed of a plastic or elastomeric sheet material (eg, polyethylene) which can be easily molded to form depressions 25,25' and cut to form openings 23,23'. The assembled device 10 is preferably water resistant (ie, splash proof) and is most preferably waterproof. The system

has a low profile that easily conforms to the body thereby allowing freedom of movement at, and around, the wearing site. The anode/drug reservoir 26 and the cathode/salt reservoir 28 are located on the skin-contacting side of device 10 and are sufficiently separated to prevent accidental electrical shorting during normal handling and use.

The device 10 adheres to the patient's body surface (eg, skin) by means of a peripheral adhesive 30 which has upper side 34 and body-contacting side 36. The adhesive side 36 has adhesive properties which assures that the device 10 remains in place on the body during normal user activity, and yet permits reasonable removal after the predetermined (eg, 24-hour) wear period. Upper adhesive side 34 adheres to lower housing 20 and retains the electrodes and drug reservoirs within housing depressions 25,25' as well as retains lower housing 20 attached to upper housing 16.

The push button switch 12 is located on the top side of device 10 and is easily actuated through clothing. A double press of the push button switch 12 within a short period of time, eg, three seconds, is preferably used to activate the device 10 for delivery of drug, thereby minimizing the likelihood of inadvertent actuation of the device 10.

Upon switch activation an audible alarm signals the start of drug delivery, at which time the circuit supplies a predetermined level of DC current to the electrodes/reservoirs for a predetermined (eg, 10 minute) delivery interval. The LED 14 remains "on" throughout the delivery interval indicating that the device 10 is in an active drug delivery mode. The battery preferably has sufficient capacity to continuously power the device 10 at the predetermined level of DC current for the entire (eg, 24 hour) wearing period.

The present invention is further explained by the following examples which are illustrative of, but do not limit the scope of, the present invention.

## EXAMPLE 1

The following experiment was conducted in order to determine the necessary minimum concentration of fentanyl salt in a donor reservoir of a transdermal electrotransport delivery device in order to ensure that the transdermal electrotransport fentanyl flux remains approximately proportional to the level of applied electrotransport current.

Anodic donor reservoir gels, having varying loadings of fentanyl HCl, were prepared having the following composition:

	<u>Material</u>	<u>(wt%)</u>
10	Water	81.3
	PVOH	15.0
	Fentanyl HCl	1.7
	Polacrilin	0.1
	0.5 N NaOH	1.9

15 The combination of Polacrilin and NaOH acted as a buffer to maintain the pH of the gels around 5.5. Polacrilin (also known as Amberlite IRP-64) is sold by Rohm & Haas of Philadelphia, PA. The materials were mixed in a beaker at elevated temperature of 90°C to 95°C, poured into foam molds and stored overnight at -35°C to cross-link the PVOH. The gels had a skin contact area of 2 cm<sup>2</sup> and a thickness of 1.6 mm. The gels had a fentanyl HCl concentration of 21 mg/mL of water. A silver foil anodic electrode was laminated to one surface of the gels.

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D1 25 The transdermal electrotransport fentanyl flux from these gels was measured by *in vitro* flux studies using a two-compartment diffusion cell and human cadaver skin. The gels were mounted on the stratum corneum side of heat stripped human cadaver epidermis taken from back skin samples.



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The other side of the epidermis was exposed to a receptor compartment, having a volume of 4 cm<sup>2</sup>, and filled with one tenth strength Dulbecco's phosphate buffered saline (pH 7.4). A counter electrode comprised of a polyisobutylene film loaded with silver chloride powder was placed in the receptor compartment.

The donor and counter electrodes were electrically connected to a galvanostat which was set to apply a constant DC current of 200  $\mu$ A (ie, 100  $\mu$ A/cm<sup>2</sup>). The current was applied continuously for 16 hours and the receptor compartment was sampled every hour over the 16 hour period.

10 Six identical flux experiments were run with different skin samples and the transdermal flux was averaged over the 6 runs. The transdermal fentanyl flux increased over the first 8 hours of current application, after which the flux remained approximately constant (ie, steady state flux was reached after 8 hours). The fentanyl concentration was estimated by  
15 subtracting the amount of fentanyl delivered through the skin into the receptor solution from the original fentanyl content in the donor gel, and dividing by the weight of water in the gel.

The normalized transdermal fentanyl flux, calculated as a percentage of the steady state transdermal flux, was plotted versus fentanyl concentration in the gel, and is shown in FIG. 2. As can be seen from FIG.  
20 2, the normalized flux remains at or near 100% at fentanyl HCl concentrations above about 6 mg/mL. The normalized flux begins to drop off as the fentanyl HCl concentration falls below 6 mg/mL and particularly below about 4 mg/mL. These results show that as the fentanyl HCl  
25 concentration falls below about 6 mg/mL, a more significant portion of the applied electrotransport current is carried by ions other than fentanyl ions and the fentanyl flux is more dependent on the fentanyl HCl concentration. Thus, to ensure a predictable fentanyl flux with a particular level of applied

electrotransport current, the fentanyl HCl concentration in the donor reservoir is preferably maintained above about 6 mg/mL.

## EXAMPLE 2

The following study was conducted to determine the amount of fentanyl hydrochloride drug loading which is necessary to prevent silver migration, resulting in transient epidermal discoloration, from a transdermal fentanyl electrotransport delivery device having a donor reservoir gel weighing about 0.6 g and having a skin contact area of about 2.8 cm<sup>2</sup>, which device is worn for a period of up to 24 hours and which applies an electrotransport current of 240  $\mu$ A (ie, a current density of 87  $\mu$ A/cm<sup>2</sup>) over a delivery interval of about 10 minutes to deliver a 40  $\mu$ g dose, and which can deliver up to 80 of such doses over the 24 hour wearing period. Thus, the device has the ability to deliver up to 3.2 mg of fentanyl (80 x 40  $\mu$ g = 3.2 mg) for therapeutic purposes.

Fentanyl HCl-containing polyvinyl alcohol (PVOH) hydrogel-based donor reservoirs, each reservoir having a total weight of about 0.15 g, were made with the following composition:

	<u>Material</u>	<u>(wt%)</u>
	Water	80.8
	PVOH	15.0
	Fentanyl HCl	2.0
	Polacrillin	0.1
	0.5 N NaOH	2.1

The materials were mixed in a jacketed beaker at 90°C and then 0.15 g aliquots of the liquid gel were dispensed into foam molds and frozen

overnight at temperatures ranging from -15 to -50°C. The gels had a disk shape with an area of 1.0 cm<sup>2</sup> and a thickness of 1.6 mm.

A silver foil was laminated to one surface of each of the gels to form an anodic donor electrode assembly comprised of the silver foil anode and the fentanyl containing gel reservoir. Counter electrode assemblies were made using similarly sized PVOH gels which contained citrate buffered saline (pH 4). A silver chloride cathodic electrode (ie, silver chloride powder-loaded polyisobutylene film) was laminated to one surface of the counter gels. The electrodes were electrically connected to custom made power sources which applied a constant DC current of 240 µA (87 µA/cm<sup>2</sup>).

The electrotransport systems were applied to the upper outer arms of six male volunteers and worn for a period of 15 hours, which is about 10% longer than the maximum time of current application from this system (ie, 80 x 10 minutes = 13.3 hrs). Over the 15 hour wearing period, the systems applied current continuously, after which the systems were removed and the arm of each subject was closely examined to determine if transient epidermal discoloration (TED), caused by migration of silver ions formed in the anodic electrode assembly, had occurred. The subjects were again examined one hour and again at 24 hours after system removal to confirm the initial TED reading. In all six subjects, no TED occurred at the site of attachment of the anodic electrode assembly. This indicates that a fentanyl HCl loading of about 1.8 to 2 wt%, or about 3 mg in these gels, provides a sufficient quantity of chloride ions to prevent migration of silver ions, formed by oxidation of the silver anode, into the skin of the patient over the 15 hour wearing period. Thus, an electrotransport system which applies the same level of electrotransport current over a maximum dosing period of 13.3 hours will likewise exhibit no TED, even under conditions of maximum usage. The 2 wt% fentanyl HCl loading in these PVOH-based donor gel reservoirs can be scaled-up to larger reservoirs. Thus, for a fentanyl HCl-containing PVOH-

based donor reservoir having a total weight of about 0.6 g, the reservoir containing substantially no other source of chloride ions other than the drug counter ions, the fentanyl HCl loading should be at least about 11 mg (ie,  $1.8 \text{ wt\%} \times 0.6 \text{ g} = 11 \text{ mg}$ ) even though the maximum amount of fentanyl which can be delivered from the device over the 24 hour wearing period is only about 3.2 mg fentanyl. Thus, in order to prevent silver migration in this device under conditions of maximum usage, an excess amount of fentanyl HCl must be loaded into the anodic donor reservoir, which excess loading is about 3 to 4 times the amount of fentanyl needed for therapeutic purposes.

### EXAMPLE 3

The following studies were conducted to determine the transdermal electrotransport dosing level required to achieve an acceptable level of analgesia in human patients suffering from moderate to severe post-operative pain. The study was conducted in 132 post-operative male and female patients who were expected to have moderate to severe pain after surgery, including orthopedic (shoulder, knee, long bone) and abdominal (urological, gynecological) surgeries. The patients wore one of two different electrotransport fentanyl HCl delivery devices on the upper arm for 24 hours following surgery. Both devices applied electrotransport current for a delivery interval of 10 minutes upon activating a push button switch on the device. The first device, worn by 79 of the 132 patients, applied an electrotransport current of 150  $\mu\text{A}$  which delivered an average fentanyl dose of 25  $\mu\text{g}$  over the 10 minute delivery interval. The second device, worn by 53 of the 132 patients, applied an electrotransport current of 240  $\mu\text{A}$  which delivered an average fentanyl dose of 40  $\mu\text{g}$  over the 10 minute delivery interval.

In both devices, the patients could self-administer up to 6 doses every hour. Patients using the first (ie, 25  $\mu\text{g}$  dose) device could apply a maximum

of 144 doses. Patients using the second (ie, 40 µg dose) device were allowed to apply up to a maximum number of 80 doses.

Both devices were two-part systems which included a reusable electronic controller and a single use/disposable drug-containing unit. Each drug unit contained an anodic fentanyl HCl-containing donor gel and a cathodic saline-containing counter gel. All gels had a skin contact area of 2 cm<sup>2</sup> and a thickness of 0.16 cm. The approximate weight of the donor gels was 350 mg. The anodic donor gels in the 25 µg dose and 40 µg dose systems were the same size and composition, only the applied electrotransport current level was different. The cathodic counter electrode assemblies each had a PVOH based gel which contained citrate buffered saline. A silver chloride cathodic electrode was laminated to one surface of the counter gel. The 25 µg and 40 µg dose anodic gels had the following composition:

	<u>Material</u>	<u>(wt%)</u>
	Water	73.2
	PVOH	10.0
	Fentanyl HCl	1.4
	Polacrilin	0.3
	Polacrilin potassium	0.1
	Glycerin	5.0
	Cholestyramine resin	10.0

All patients were initially titrated to an acceptable level of analgesia with intravenous (IV) fentanyl in the recovery room immediately following surgery. Within 3 hours after surgery when the patients had met the usual institutional standards for discharge from the recovery room and were able to operate their worn electrotransport delivery device, the patients were moved

to a ward where they could self administer fentanyl by transdermal electrotransport for the management of their pain. In the event the electrotransport fentanyl delivery regimen was insufficient to control pain, the patients were retitrated with supplemental fentanyl through IV administration to achieve adequate analgesia.

In the 25 µg dose group, 38 of 79 patients (ie, 48% ) required no supplemental IV fentanyl after leaving the recovery room. In the 40 µg dose group, 47 of 53 patients (ie, 89%) required no supplemental IV fentanyl after leaving the recovery room. Based on these percentages, it was determined that the 25 µg dose regimen was insufficient, and the 40 µg dose regimen was sufficient, to treat the pain associated with these types of surgical procedures in a majority of the patients tested. Based on the fact that the 25 µg dose regimen was analgesically effective for about half the patients, it is likely that this lower dose would be effective in treating less severe acute pain such as that experienced with hernia repair, kidney stones, arthritis pain, laparoscopic procedures, and other conditions involving less severe pain than that associated with major surgeries.

Pain intensity was assessed at baseline immediately before activation of the first on-demand dose and again at times 0.5, 1, 2, 3, 4, 6, 8, 12, 16, 20 and 24 hours after the devices were first activated. The patients were asked to assess pain intensity by marking on a 10 cm long strip, containing a scale of 1 to 100, with 1 being associated with no pain and 100 being associated with the most severe intensity pain. The quality of analgesia was evaluated by a categorical rating of excellent, good, fair or unsatisfactory according to the same time schedule as that for the pain intensity measurements.

The quality of analgesia and pain intensity data for the 53 patients using the 40 µg dose electrotransport devices are shown in FIGS. 3 and 4, respectively.

5 Skin sites beneath the anode and cathode gels were assessed at 1, 6 and 24 hours following removal of the devices and evaluated for topical (eg, irritation) effects. The topical effects data are shown in Table 1.

TABLE 1

	Hours Post Removal	ETS Skin Site	Score	Edema (%)	Erythema (%)	Extent of Erythema (%)	Itching (%)	Papules (%)	Pustules (%)
5									
	1	Anode	0	74	15	19	91	92	100
			1	8	49	32	6	6	0
			2	19	36	49	4	2	0
10		Cathode	0	92	72	74	94	94	100
			1	6	19	13	4	6	0
			2	2	9	13	2	0	0
	6	Anode	0	74	15	17	89	92	100
			1	11	43	34	8	8	0
15			2	15	40	49	4	0	0
			3	0	2	0	0	0	0
		Cathode	0	92	68	68	91	91	100
			1	4	19	13	9	6	0
			2	4	9	19	0	4	0
20			3	0	4	0	0	0	0
	24	Anode	0	83	34	36	91	96	98
			1	9	40	38	8	4	2
			2	8	26	36	2	0	0
			3	0	0	0	0	0	0
		Cathode	0	91	70	70	91	89	98
25			1	6	19	15	8	8	0
			2	4	8	15	2	4	2
			3	0	4	0	0	0	0
30	Erythema: 0 = None 1 = Barely perceptible redness 2 = Definite redness 3 = "Beet" redness				Itching: 0 = None 1 = Mild 2 = Moderate 3 = Severe				
35	Edema, Papules, Pustules, Extent of Erythema:				0 = None 1 = <50% of occluded area 2 = >50% of occluded area				



## EXAMPLE 4

Two fentanyl hydrochloride-containing anodic donor reservoir PVOH-based gels were made having the following compositions:

## Donor Gel Formulations:

5	<u>Material</u>	<u>wt %</u>	<u>wt %</u>
	Purified Water	86.3	85.3
	Washed PVOH	12.0	12.0
	Fentanyl HCL	1.7	1.7
	Hydroxy Methylcellulose	----	1.0

10 With both formulations, the water and PVOH are mixed at a temperature between 92°C and 98°C followed by the addition of fentanyl hydrochloride and subsequent further mixing. The liquid gel was then pumped into foam molds having a disc-shaped cavity. The molds were placed in a freezer overnight at -35°C to cross-link the PVOH. The gels can  
15 be used as anodic donor reservoirs suitable for transdermal electrotransport fentanyl delivery.

In summary, the present invention provides a method for improving the transdermal electrotransport of water soluble salts of fentanyl, and sufentanil which are preferably delivered from an electrotransport device  
20 having a silver anodic donor electrode and a hydrogel based donor reservoir. The electrotransport device is preferably a patient-controlled device. The hydrogel formulation contains a drug concentration which is sufficient to maintain transdermal electrotransport drug flux for a predetermined current level, to inhibit silver ion migration to the skin of a wearer of the  
25 electrotransport device and thus, prevent transient epidermal discoloration, and to provide an acceptable level of analgesia.